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Jingyu Liang, and David G. I. Kingston

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TWO NEW TAXANE DITERPENOIDS FROM *TAXUS MAIREI*JINGYU LIANG¹ and DAVID G.I. KINGSTON*Department of Chemistry, Virginia Polytechnic Institute and State University,
Blacksburg, Virginia 24061-0212

ABSTRACT.—Two new taxane diterpenoids were isolated from the heartwood of *Taxus mairei* and identified as 5 α ,7 β ,9 α ,10 β ,13 α -pentaacetoxy-2 α -benzoyloxy-4 α ,20-dihydroxytax-11-ene [**2**] and 7 β ,9 α ,10 β ,13 α ,20-pentaacetoxy-2 α -benzoyloxy-4 α ,5 α -dihydroxytax-11-ene [**3**]. The structures of **2** and **3** were determined primarily on the basis of analysis of their ¹H-nmr, ¹³C-nmr, ¹H-¹H 2D COSY, ¹H-¹³C 2D COSY, DEPT, and mass spectra. This is the first report of naturally occurring taxoids lacking a D ring but oxygenated at the 4, 5, and 20 positions.

The discovery that the diterpenoid natural product taxol [**1**] (1) is clinically active against ovarian and breast cancer (2) has stimulated a renewed interest in the isolation of related compounds from various *Taxus* species. These compounds could in some cases serve as potential sources of taxol through appropriate chemical transformations, but they can also serve as sources for taxol analogues, thus extending our understanding of structure-activity relationships in this area. A recent review (3) lists over 100 taxane diterpenoids (or taxoids), and new ones are being described on a regular basis.

Previous studies on the diterpenoids in the bark and the heartwood of *Taxus mairei* (Lemee et Levl.) S.Y. Hu (Taxaceae) have resulted in the isolation of various diterpenoids (4,5) and many new and known taxoids (6–13). In a continuation of the investigation of *T. mairei* by one of us (4–8), we now report the isolation and structure elucidation of two new taxoids from the heartwood of this plant.

Extraction of the heartwood of *T. mairei* with EtOH, followed by extraction of the dried EtOH extract with CHCl₃, gave a residue which was purified by cc on Si gel and by preparative tlc to yield the two new taxoids **2** and **3**.

Compound **2** had the composition C₃₇H₄₈O₁₄ as deduced by a combination of low-resolution ms and ¹³C-nmr spectroscopy. Its ¹H-nmr spectrum (Table 1) showed the presence of four C-methyl groups and five acetate methyl groups, together with one benzoyl group. The presence of a taxene skeleton was inferred from the observation of characteristic resonances, such as the broad singlet at 2.30 ppm for the C-18 methyl group, and from the source of the material; this conclusion was corroborated by detailed analysis of the ¹H- and ¹³C-nmr spectra (Table 1).

A one-proton multiplet at 1.92 ppm was correlated by ¹H-¹H COSY with protons resonating at 5.82 and 2.66 ppm, and was thus assigned to C-1. The proton at 5.82 ppm, assigned to C-2, correlated with one at 3.12 ppm which must therefore be C-3. The C-3 proton appeared as a doublet (*J*=4.9 Hz), indicating that the C-4 position is fully substituted. The C-5 proton appeared as a broad singlet at 5.21 ppm, and it could be correlated with a signal at 1.94, assigned to one of the C-6 protons. The other C-6 proton is assigned to a multiplet at 2.65 ppm, but the correlations of this proton were complicated by overlapping signals, and the assignment remains tentative. The C-7 proton was assigned to a doublet of doublets at 5.55 ppm, correlating with the C-6 proton at 1.94 ppm. The C-9 and C-10 protons appeared as a pair of doublets at 5.97 and 6.19 ppm, respectively, and the C-13 proton appeared as a multiplet at 5.80 ppm. This signal correlated with the C-14 protons at 2.41 and 2.66 ppm. An isolated spin system with signals at 3.43 and 3.71 ppm was assigned to C-20. The ¹³C-nmr spectrum of **2**

¹Present address: Department of Phytochemistry, China Pharmaceutical University, 24 Tong Jia Xiang, Nanjing, People's Republic of China.

TABLE 1. ^{13}C - and ^1H -nmr Spectra of $5\alpha,7\beta,9\alpha,10\beta,13\alpha$ -Pentaacetoxy- 2α -benzoyloxy- $4\alpha,20$ -dihydroxytax- 11 -ene [2].

Position	^{13}C	Carbon type ^a	$^1\text{H}^b$	^1H - ^{13}C COSY ^c	^1H - ^1H COSY
1	48.08	D	1.92 (1H, m)	*	C-2, C-14
2	71.99	D	5.82 (1H, m)	*	C-3, C-1
3	45.87	D	3.12 (1H, d, 4.9)	*	C-2
4	75.42	S	—		
5	73.00	D	5.21 (1H, t, $J=2$)	*	C-6
6	30.90	T	194 (1H, m) 2.65 (1H, m)		C-7, C-5 C-6
7	68.62	D	5.55 (1H, dd, 11.8, 4.8)	*	
8	46.07	S	—		
9	75.20	D	5.97 (1H, d, 11.0)	*	C-10
10	71.59	D	6.19 (1H, d, 11.0)	*	C-9
11	134.80	S	—		
12	137.98	S	—		
13	70.71	D	5.80 (1H, m)	*	C-14 α , C-14 β
14	27.67	T	2.41 (1H, dd, 15.7, 5.2, αH) 2.66 (1H, ddd, 15.7, 8.7, 8.6, βH)		C-13, C-14 β C-13, C-1, C-14 α
15	38.04	S	—		
16	26.04	Q	1.81 (3H, s)	*	
17	31.91	Q	1.07 (3H, s)	*	
18	15.50	Q	2.30 (3H, brs)		
19	14.48	Q	1.00 (3H, s)	*	
20	64.48	T	3.43 (1H, d, 11.1) 3.71 (1H, d, 11.1)		C-20a C-20b
OAc	20.90	Q	1.99 (3H, s)		
	169.24	S			
OAc	21.09	Q	2.04 (3H, s)		
	169.76	S			
OAc	21.55	Q	2.00 (3H, s)		
	170.23	S			
OAc	21.81	Q	2.32 (3H, s)		
	170.80	S			
OAc	22.03	Q	2.33 (3H, s)		
	172.50	S			
OCOC ₆ H ₅ , ...	165.32	S	—		
i	129.85	S	—		
o	129.89	D	8.07 (2H, d, 7.4)	*	meta-Ar
m	128.82	D	7.46 (2H, m)	*	ortho-Ar
p	133.66	D	7.59 (1H, m)	*	

^aS=C, D=CH, T=CH₂, Q=Me. Assignments made by the ADEPT technique.

^bMultiplicity and coupling constant (s) in Hz in parentheses.

^cAn asterisk indicates that a ^1H - ^{13}C correlation was observed in a HETCOR spectrum.

Hz, comparable to values observed for these protons in molecules such as baccatin VI [5] (3). Finally, the C-13 acetoxy group is assigned an α configuration on the basis of analogy with all the other derivatives of this type that have been isolated. Based on the above data, the new taxoid is assigned the structure and stereochemistry as illustrated in 2.

The second compound obtained was isomeric with 2 and had similar ^1H - and ^{13}C -nmr spectra (Table 2). The major difference between the two compounds was in the

TABLE 2. ^1H - and ^{13}C -nmr Spectra of $7\beta,9\alpha,10\beta,13\alpha,20$ -Pentaacetoxy- 2α -benzoyloxy- $4\alpha,5\alpha$ -dihydroxytax-11-ene [3].

Position	^{13}C	Carbon type ^a	$^1\text{H}^b$	^1H - ^{13}C COSY ^c	^1H - ^1H COSY
1	48.04	D	1.87 (1H, m)	*	C-14 α , C-2
2	72.39	D	5.89 (1H, m)	*	C-1, C-3
3	43.31	D	3.31 (1H, d, 4.9)	*	C-2
4	75.80	S	—		
5	70.70	D	3.70 (1H, br.s)	*	C-6
6	31.46	T	1.80 (1H, m)		C-7
			1.95 (1H, m)		C-7, C-5
7	68.39 ^d	D	5.68 (1H, dd, 11.8, 4.8)	*	C-6 α , C-6 β
8	46.20	S	—		
9	75.48	D	5.87 (1H, d, 10.9)	*	C-10
10	72.13	D	6.23 (1H, d, 11.0)	*	C-9
11	134.47	S	—		
12	139.20	S	—		
13	69.99 ^d	D	5.68 (1H, m)	*	C-18, C-14
14	27.52	T	2.55 (1H, dd, 15.5, 4.4, αH)		C-1
			2.65 (1H, ddd, 15.5, 7.8, 3.9, βH)		C-13, C-1,
15	37.72	S	—		
16	25.91	Q	1.75 (3H, s)	*	
17	32.50	Q	1.01 (3H, s)	*	
18	15.94	Q	2.26 (3H, br.s)	*	C-13
19	14.52	Q	1.03 (3H, s)	*	
20	65.43	T	4.02 (1H, d, 11.9)		C-20'
			4.23 (1H, d, 11.9)		C-20
OAc	20.45	Q	1.78 (3H, s)	*	
	169.18	S			
OAc	20.92	Q	2.04 (3H, s)	*	
	169.51	S			
OAc	21.11	Q	1.98 (3H, s)	*	
	169.99	S			
OAc	21.27	Q	2.15 (3H, s)	*	
	170.08	S			
OAc	21.56	Q	2.05 (3H, s)	*	
	170.23	S			
OCOC ₆ H ₅	165.13	S	—		
C ₆ H ₅ , i	129.80	S	—		
o	129.90	D	7.99 (2H, d, 7.0)	*	<i>m</i> -C ₆ H ₅
m	128.74	D	7.47 (2H, m)	*	<i>o</i> -C ₆ H ₅
p	133.64	D	7.60 (1H, m)	*	

^aS=C, D=CH, T=CH₂, Q=Me. Assignments made by the ADEPT technique.

^bMultiplicity and coupling constant (s) in Hz in parentheses.

^cAn asterisk indicates that a ^1H - ^{13}C correlation was observed in a HETCOR spectrum.

^dThese signals could not be distinguished by the HETCOR spectrum; assignment is based on analogy to compound 2.

chemical shifts of the C-5 and C-20 protons. In compound 2 these were at 5.21 and 3.43/3.71 ppm, respectively, while in compound 3 they occurred at 3.70 and 4.02/4.23 ppm. These chemical shift changes clearly indicate that compound 3 has a free hydroxyl group at C-5 and an acetoxyl group at C-20. The structure and stereochemistry represented by 3 are thus assigned to this second compound.

Compounds **2** and **3** are the first naturally occurring examples of a taxoid oxygenated at C-4, C-5, and C-20 but lacking a D ring. Compounds of this type have previously been formed by opening of the oxetane ring of taxol (14,15) and have also been prepared as intermediates in a synthesis of a baccatin IV derivative from a taxicin-I derivative (16). Compounds of this type have also been proposed as intermediates in the biosynthesis of the oxetane ring system of baccatin III and taxol (17), and the isolation of compounds **2** and **3** thus provides additional support for this proposal.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Nmr spectra were obtained on a Varian Unity 400 spectrometer in CDCl₃ at ambient temperature. ¹H chemical shifts are recorded in ppm from internal TMS, and ¹³C shifts are based on the CHCl₃ signal at 77.0 ppm. Ms were obtained on a VG 7070 mass spectrometer, using the ci method. Ir spectra were recorded as KBr pellets on a Perkin-Elmer 283B spectrometer, and uv spectra were recorded on a Perkin-Elmer 330 spectrometer. Optical rotations were determined on a Perkin-Elmer 141 polarimeter. Melting points were determined on a Thermolyne hot stage and are uncorrected.

ISOLATION OF COMPOUNDS 2 AND 3.—Plant material was collected in Fujian Province, People's Republic of China, and a voucher specimen is deposited in the Department of Phytochemistry, China Pharmaceutical University. A 16.6 kg sample of heartwood of *T. mairei* was air-dried, ground, and extracted with EtOH. The EtOH extract was evaporated to dryness and then re-extracted with CHCl₃ in a Soxhlet extractor to give 60 g of CHCl₃-soluble material. This material was subjected to twofold chromatography on Si gel with elution by CH₂Cl₂/hexanes. After the second separation the fractions that eluted immediately after 1-dehydroxybaccatin VI were subjected to preparative tlc [Si gel F₂₅₄, 20×20 cm, EtOAc-hexanes (5:3)] to give **2** (17.6 mg) and **3** (46.7 mg).

5 α ,7 β ,9 α ,10 β ,13 α -Pentaacetoxy-2 α -benzoyloxy-4 α ,20-dihydroxytax-11-ene [**2**].—Mp 114–115° (Me₂CO); [α]_D²⁰ –30° (c =0.1, CHCl₃); uv λ max (EtOH) 228 (ϵ 4643), 270 (542); ir (KBr) ν max 3400, 1720, 1705, 1360, 1200, 1020, 700 cm⁻¹; ¹H and ¹³C nmr see Table 1; cims m/z [MH–H₂O]⁺ 699 (2), [MH–HOAc]⁺ 657 (10), [MH–H₂O–HOAc]⁺ 639 (10), 597 (15), 579 (5) 535 (25), 517 (15), 492 (10), 475 (25), 457 (15), 433 (15), 415 (50), 347 (30), 373 (30), 355 (60), 337 (40), 313 (35), 295 (80), 123 (80), 105 (100).

7 β ,9 α ,10 β ,13 α ,20-Pentaacetoxy-2 α -benzoyloxy-4 α ,5 α -dihydroxytax-11-ene [**3**].—Mp 122–123° (Me₂CO); [α]_D²⁰ –37° (c =0.1, CHCl₃); uv λ max (EtOH) 228 (6386), 270 (1214); ir (KBr) 3400, 1720, 1705, 1360, 1220, 1010, 700 cm⁻¹; ¹H and ¹³C nmr see Table 2; cims m/z [MH–H₂O] 699 (3), [MH–HOAc]⁺ 657 (18), [MH–H₂O–HOAc]⁺ 639 (15), 597 (15), 579 (10), 535 (25), 517 (15), 492 (10), 475 (30), 457 (15), 433 (12), 415 (60), 347 (40), 373 (25), 355 (70), 337 (45), 313 (30), 295 (100), 277 (100), 123 (40), 105 (60).

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LITERATURE CITED

1. M.C. Wani, H.L. Taylor, M.E. Wall, P. Coggon, and T.T. McPhail, *J. Am. Chem. Soc.*, **93**, 2325 (1971).
2. E.K. Rowinsky and R.C. Donehower, *J. Natl. Cancer Inst.*, **83**, 1778 (1991).
3. D.G.I. Kingston, A.A. Molinero, and J.M. Rimoldi, in: "Progress in the Chemistry of Organic Natural Products." Ed. by W. Herz, G.W. Kirby, and Ch. Tamm, Springer-Verlag, New York, 1993, Vol. 59, pp. 1–188.
4. J. Liang, Z. Min, T. Tanaka, M. Mizuno, and M. Iinuma, *Acta Chim. Sin. (Huaxue Xuebao)*, **46**, 21 (1988).
5. J. Liang, Z. Min, M. Iinuma, T. Tanaka, and M. Mizuno, *Chem. Pharm. Bull.*, **35**, 2613 (1987).
6. J. Liang, Z. Min, and M. Niwa, *Huaxue Xuebao*, **46**, 1053 (1988).
7. J. Liang, Z. Min, and M. Mizuno, T. Tanaka, and M. Iinuma, *Phytochemistry*, **27**, 3674 (1988).
8. Z.D. Min, H. Jiang, and J. Liang, *Yaoxue Xuebao*, **24**, 673 (1989).
9. L.C. Chuang, K.J. Chen, Y.S. Lin, and F.C. Chen, *Huaxue*, **48**, 275 (1990).
10. C.L. Liu, Y.C. Lin, Y.M. Lin, and F.C. Chen, *T'ai-wan K'o Hsueh*, **38**, 119 (1984).
11. M.K. Yeh, J.S. Wang, L.P. Liu, and F.C. Chen, *J. Chin. Chem. Soc.*, **35**, 309 (1988).

12. M.K. Yeh, J.S. Wang, W.L. Yang, and F.C. Chen, *Proc. Natl. Sci. Counc., Rep. China, Part A: Phys. Sci. Eng.*, **12**, 89 (1988).
13. M.K. Yeh, J.S. Wang, L.P. Liu, and F.C. Chen, *Phytochemistry*, **27**, 1534 (1988).
14. G. Samaranayake, N.F. Magri, C. Jitrangsi, and D.G.I. Kingston, *J. Org. Chem.*, **56**, 5114 (1991).
15. A. Wahl, F. Guéritte-Voegelein, D. Guénard, M.-T. LeGoff, and P. Potier, *Tetrahedron* **48**, 6965 (1992).
16. L. Ettouati, A. Ahond, C. Poupat, and P. Potier, *Tetrahedron*, **47**, 9823 (1991).
17. D.P. Della Casa de Marcano, T.G. Halsall, E. Castellano, and O.J.R. Hodder, *J. Chem. Soc. D (Chem. Commun.)* 1382 (1970).

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